

Cambridge International AS & A Level

PHYSICS

Paper 2 AS Level Structured Questions MARK SCHEME Maximum Mark: 60 9702/22 February/March 2025

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This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the February/March 2025 series for most Cambridge IGCSE, Cambridge International A and AS Level components, and some Cambridge O Level components.

Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptions for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

- 1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
- 2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
- 3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
- 4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
- 5 <u>'List rule' guidance</u>

For questions that require *n* responses (e.g. State **two** reasons ...):

- The response should be read as continuous prose, even when numbered answer spaces are provided.
- Any response marked *ignore* in the mark scheme should not count towards **n**.
- Incorrect responses should not be awarded credit but will still count towards *n*.
- Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
- Non-contradictory responses after the first *n* responses may be ignored even if they include incorrect science.

6 <u>Calculation specific guidance</u>

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 <u>Guidance for chemical equations</u>

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Annotations guidance for centres

Examiners use a system of annotations as a shorthand for communicating their marking decisions to one another. Examiners are trained during the standardisation process on how and when to use annotations. The purpose of annotations is to inform the standard isation and monitoring processes and guide the supervising examiners when they are checking the work of examiners within their team. The meaning of annotations and how they are used is specific to each component and is understood by all examiners who mark the component.

We publish annotations in our mark schemes to help centres understand the annotations they may see on copies of scripts. Note that there may not be a direct correlation between the number of annotations on a script and the mark awarded. Similarly, the use of an annotation may not be an indication of the quality of the response.

The annotations listed below were available to examiners marking this component in this series.

Annotations

Annotation	Meaning
	information missing or insufficient for credit
AE	arithmetic error
BOD	benefit of the doubt given
CON	contradiction in response, mark not awarded
×	incorrect point or mark not awarded
ECF	error carried forward applied
I	ignore the response

Annotation	Meaning
MO	mandatory mark not awarded
POT	power of ten error
SEEN	blank page seen
SF	error in number of significant figures
TE	transcription error
~	correct point or mark awarded
ХР	incorrect physics

Question	Answer	Marks
1(a)	how close the (measured) value is to the true value (of the quantity)	A1
1(b)(i)	$(\rho =) m / V$	C1
	= $31.3 \times 10^{-3} / (1.53 \times 10^{-2})^3$ = 8.7×10^3 (kg m ⁻³)	A1
1(b)(ii)	(0.5/31.3) or (0.01/1.53)	C1
	% uncertainty = $(0.016 \times 100) + 3 \times (0.0065 \times 100)$	
	= 4%	A1
1(b)(iii)	The ranges (of the densities of A and B) overlap or the (calculated) density of A is within the range of the density of B or the difference in densities is within the uncertainty (of B)	M1
	(so) they could be the same	A1

Question	Answer	Marks
2(a)	in (rotational) equilibrium	B1
	sum / total of CW moments about a point = sum / total of ACW moments about the (same) point.	B1
2(b)(i)	The magnitudes of the moments about A are: (1700 \times 3.0) (660 x) (1300 \times 5)	C1
	Correct magnitude of any one moment about A.	
	Correct magnitudes of a second moment about A.	C1
	$(1700 \times 3.0) + (660 x) = (1300 \times 5)$	A1
	<i>x</i> = 2.1 m	

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Question	Answer	Marks
2(b)(ii)	(Upthrust =) 11 × 9.81 + 1300 = 1400 (N)	A1
2(b)(iii)	$y = \text{Upthrust} / \rho g A$	C1
	= 1400 / (990 × 9.81 × π × (0.78 / 2) ²)	
	= 0.30 m	A1
	OR	(C1)
	$[P = F / A = 1400 / \pi \times (0.78 / 2)^2 = 2930]$ y = P / ρg = 2930 / (990 × 9.81)	
	= 0.30 m	(A1)
2(b)(iv)	line with a non-zero value of depth at distance = 0	B1
	A line from distance = 0 to distance = 6.0 m with a gradient that is always positive	B1

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Question	Answer	Marks
3(a)(i)	$a = (v^2 - u^2)/2s$	C1
	$= (22^2 - 13^2) / (2 \times 180)$	
	$= 0.88 \text{ m s}^{-2}$	A1
	OR [t = $(180 \times 2) / (22 + 13) = 10.3$] $180 = 13 \times 10.3 + \frac{1}{2} a \times 10.3^2$ or $180 = 22 \times 10.3 - \frac{1}{2} a \times 10.3^2$ or $22 = 13 + a \times 10.3$	(C1)
	<i>a</i> = 0.88 m s ⁻²	(A1)
3(a)(ii)	$(\Delta)E = \frac{1}{2}m(\Delta)v^2$	C1
	gain in KE = $\frac{1}{2}m(v^2 - u^2)$	C1
	$= \frac{1}{2} \times 9400 \times (22^2 - 13^2)$	
	$= 1.5 \times 10^{6} \mathrm{J}$	A1
	OR	(C1)
	W = Fs = $ma \times d$	
	gain in KE = $9400 \times 0.88 \times 180$	(C1)
	$= 1.5 \times 10^{6} \mathrm{J}$	(A1)
3(b)(i)	rate of change of momentum	B1
3(b)(ii)	(Force =) $(2.5 \times 10^4 - 21 \times 10^4) / 15 = -1.2 \times 10^4 (N)$	A1

Question	Answer	Marks
3(b)(iii)	The change of momentum (for S and R to come to rest) is the same	B1
	(Average) force on R (to come to rest) is $(-21 \times 10^4 / 23 =) 0.91 \times 10^4 \text{ N or}$ (Average) force on R (to come to rest) is less than the force on S / less than F	B1
	(S will come to rest in) less time (than R).	B1
	OR	(B1)
	The change of momentum (for S and R to come to rest) is the same	
	Time for S to come to rest is $(-21 \times 10^4 / 1.2 \times 10^4 =)$ 17.5 s (and time for R to come to rest is 23 s)	(B1)
	(S comes to rest in) less time (than R).	(B1)

Question	Answer	Marks
4(a)	(micro)wave (from the transmitter) reflects (at metal/sheet)	B1
	The incident and reflected waves superpose	B1
	(resultant) amplitude is maximum at an antinode	B1
	(resultant) amplitude is minimum/zero at a node	B1
4(b)(i)	$\lambda = c/f$ = 3 × 10 ⁸ /6.3 × 10 ⁹	C1
	= 0.048 m	A1
4(b)(ii)	distance PQ = $\frac{1}{4}\lambda$ = 0.048 / 4 = 0.012 m	A1

Question	Answer	Marks
4(b)(iii)	 (Distance QR is the) same (as PQ) and one of: Distance (between maxima / minima) does not depend on intensity distance depends <u>only</u> on wavelength wavelength is unchanged / constant 	B1

Question	Answer	Marks
5(a)	$T = 5.8 \times 5.0 \times 10^{-4} \\ = 2.9 \times 10^{-3}$	C1
	$\lambda = vT$ or $v = f\lambda$ and $f = 1 / T$	C1
	λ = 330 × 2.9 × 10 ⁻³ or λ = 330 / 345	A1
	= 0.96 m	
5(b)	(loudspeaker) moves away (from the microphone)	B1
	at an increasing speed / whilst accelerating	B1

Question	Answer	Marks
6(a)	V = IR = 0.85 × 3.3 × 10 ⁻³	C1
	= 2.8× 10 ⁻³ V	A1
6(b)(i)	$(A =) \rho L / R$	C1
	= $1.8 \times 10^{-8} \times 0.24 / 3.3 \times 10^{-3}$ = 1.3×10^{-6} (m ²)	A1
6(b)(ii)	(n =) $2.6 \times 10^{22} / (1.3 \times 10^{-6} \times 0.24) = 8.3 \times 10^{28} (m^{-3})$	A1

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Question	Answer	Marks
6(b)(iii)	v = I/nAq	C1
	= 0.85 / (8.3 \times 10 ²⁸ \times 1.3 \times 10 ⁻⁶ \times 1.6 \times 10 ⁻¹⁹)	
	$= 4.9 \times 10^{-5} \mathrm{ms^{-1}}$	A1
	OR	(C1)
	v = IL/Nq	
	= $0.85 \times 0.24 / (2.6 \times 10^{22} \times 1.6 \times 10^{-19})$	
	$= 4.9 \times 10^{-5} \mathrm{ms^{-1}}$	(A1)
6(c)(i)	Length (of Q) is greater (than P)	B1
	(Average cross-sectional) area (of Q) is less (than P)	B1
	Resistance is proportional to length / (cross-sectional) area	M1
	(so) the resistance (of Q) is greater (than P)	A1
6(c)(ii)	A line starting from a non-zero value of drift speed at distance = 0	B1
	A line with an increasing positive gradient	B1

Question	Answer	Marks
7(a)	nucleon number of $Q = 226$ and proton number of $R = 86$	B1
7(b)	$4(u) \times 1.5 \times 10^7 \text{ or } 222(u) \times v$	C1
	$v = 4(u) \times 1.5 \times 10^7 / 222(u)$	C1
	$= 2.7 \times 10^5 \mathrm{ms^{-1}}$	A1

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Question	Answer	Marks
7(c)	Any three from: • momentum • charge • nucleon number • neutron number • proton number	Β3